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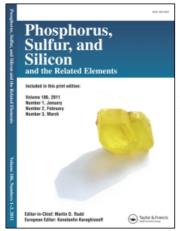
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New Ring Cleavage Reactions and Recyclizations Starting from 1,2,3-Thiadiazoles

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Keywords: 1,2,3-thiadiazoles; alkynes; ring cleavage; cyclization; benzofurans

4-Monosubstituted 1,2,3-thiadiazoles are known to decompose in the presence of base to afford the alkynethiolates. When a suitable nucleophilic function is present, a cyclization can occur. Starting from 4-(o-hydroxyphenyl)-1,2,3-thiadiazole 1, benzofurans 4 were obtained after reaction with an alkylating agent. [1] In an H-NMR spectroscopic study, we observed that both the phenolate 2 and the alkynethiolate 3 are intermediates in this reaction. Dihydroxyphenyl or hydroxynaphthyl substituted 1,2,3-thiadiazole derivatives were also converted to the corresponding hydroxybenzofuran and naphthothiophene systems 5 and 6, respectively.

Disubstituted 1,2,3-thiadiazoles 7 (R = Cl) are attacked by organometallic (Met = Li, MgBr) reagents at the sulfur atom with loss of nitrogen. When a leaving group is present, the result is an alkyne sulfide 8. Alkoxide, phenoxide or thiophenoxide bases substitute at the carbon and leave the 1,2,3-thiadiazole system of 7 (R = OR', SR') intact. The latter products afford the same alkyne sulfides 8 when treated with n-butyllithium. The alternative loss of butylsulfanyl anion does not occur. This seems to

indicate that the ring cleavage process occurs concertedly with the leaving of the 5-substituent.^[2]

$$\begin{array}{ccc}
N & R & RMet & R-C \equiv C-SR' & + N_2 \\
7 & 8 & & & & \\
X = CLOR'', SR''
\end{array}$$

When connected to a porphyrin moiety, 4-monosubstituted 1,2,3-thiadiazoles 9 will not give the expected alkynethiolate on reaction with base, but instead lose sulfur and nitrogen simultaneously to afford the alkynes 10. The acidity of the thiadiazole H-5 is lowered significantly by conjugation to the porphyrin, which is in the dianion form, allowing this alternative process. This unexpected result constitutes a new method for the generation of interesting meso-ethynyl substituted porphyrins.^[3]

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